# Worksheet

1. Volumes of 0.1 M HCl needed to neutralise 10 mL of NaOH analyte solution:

Run 1:

Run 2:

Run 3:

1. Using your result from run 3, work out the concentration of NaOH in the analyte solution:
2. Observations at the endpoint:
3. The actual concentration of NaOH is 0.2M, what do your results tell you about the indicators being used?